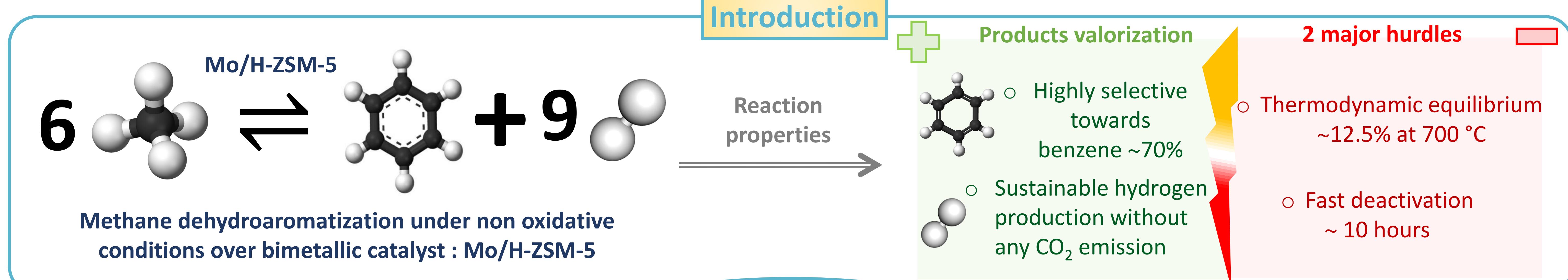


On the true nature of deactivation mode in methane dehydroaromatization reaction

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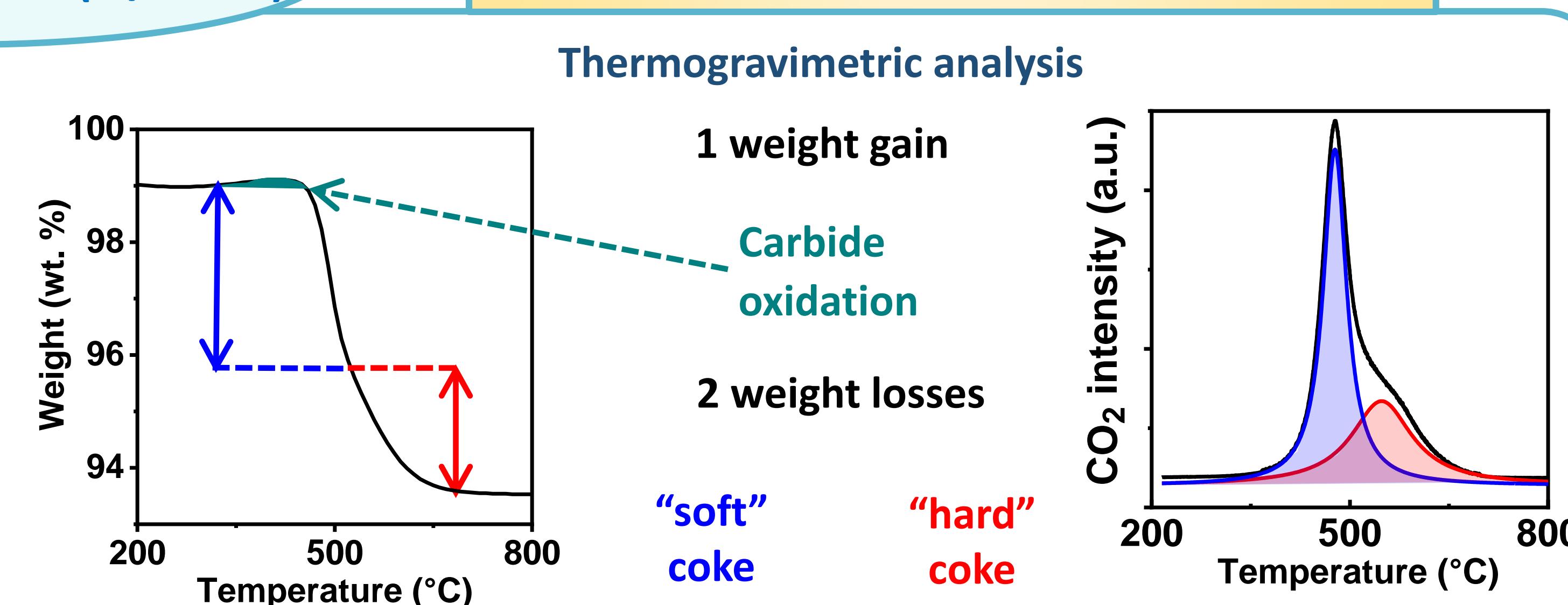
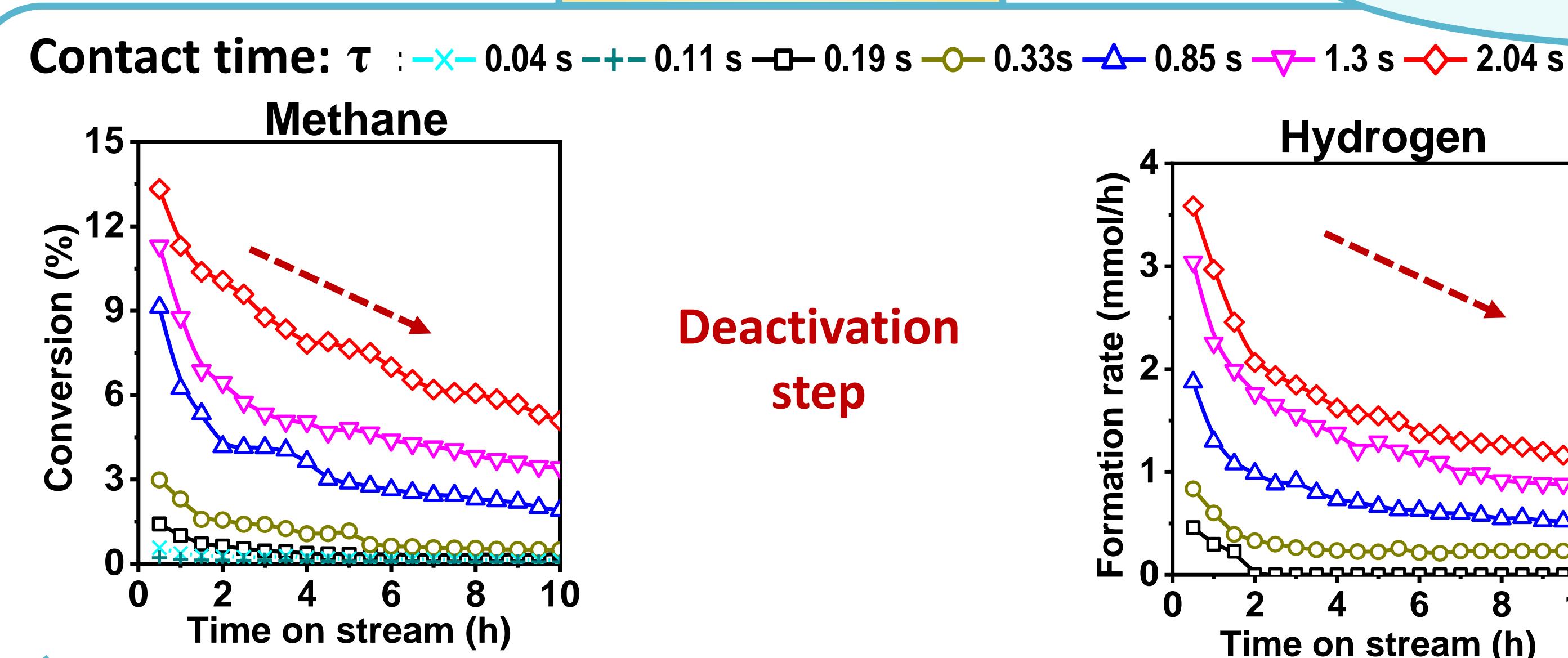
^b Univ. Lille, CNRS, Centrale Lille, Univ. Artois, UMR 8181–UCCS–Unité de Catalyse et Chimie du Solide, F-59000 Lille, France



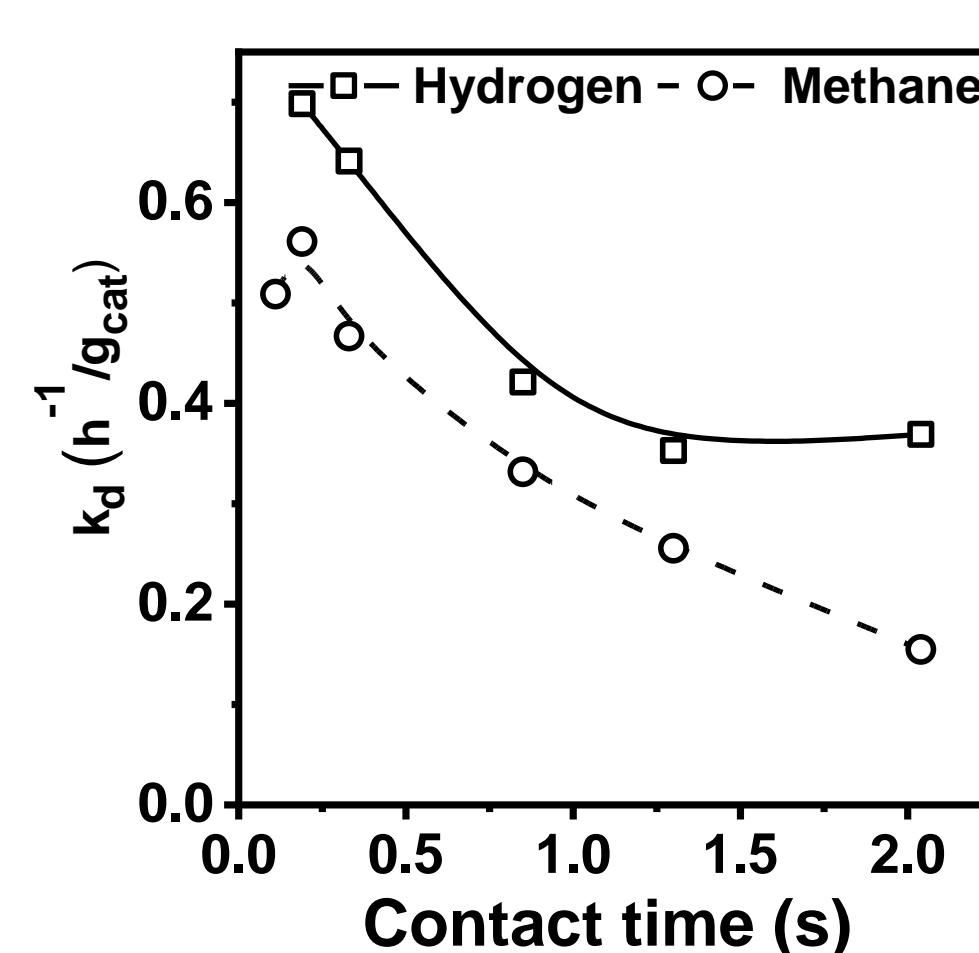
Catalytic tests

Fresh catalyst
3wt.%Mo/H-ZSM-5 (Si/Al=25)

Spent catalysts characterisation

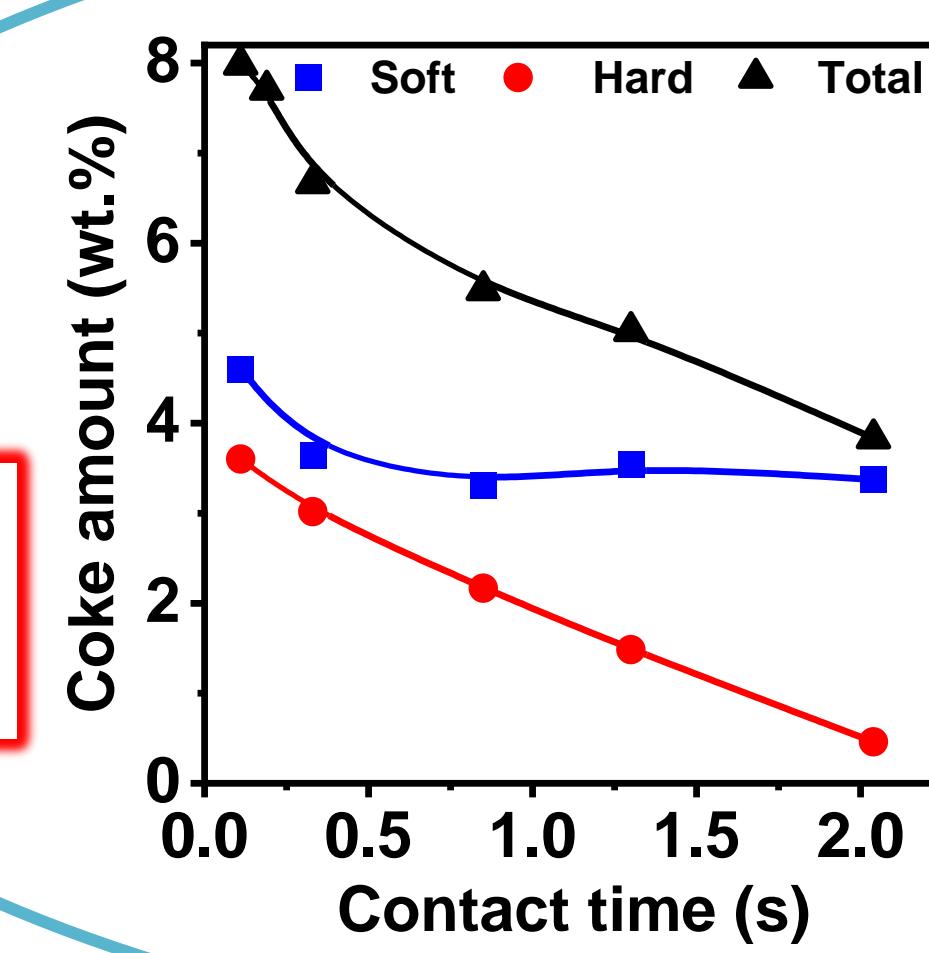


K_d calculated considering a pseudokinetic order of 1



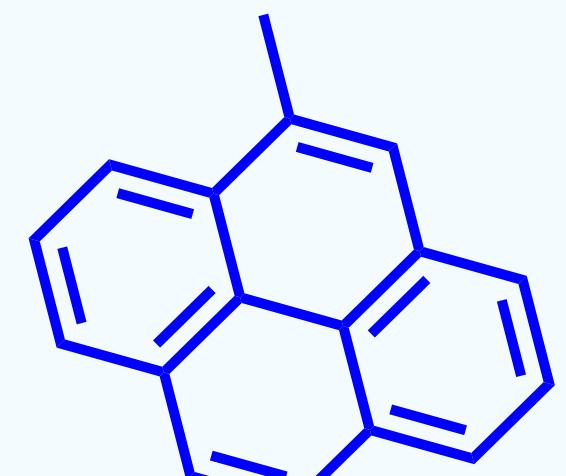
Unconventional behavior

Contact time increases → Lower coke amount Better catalyst stability

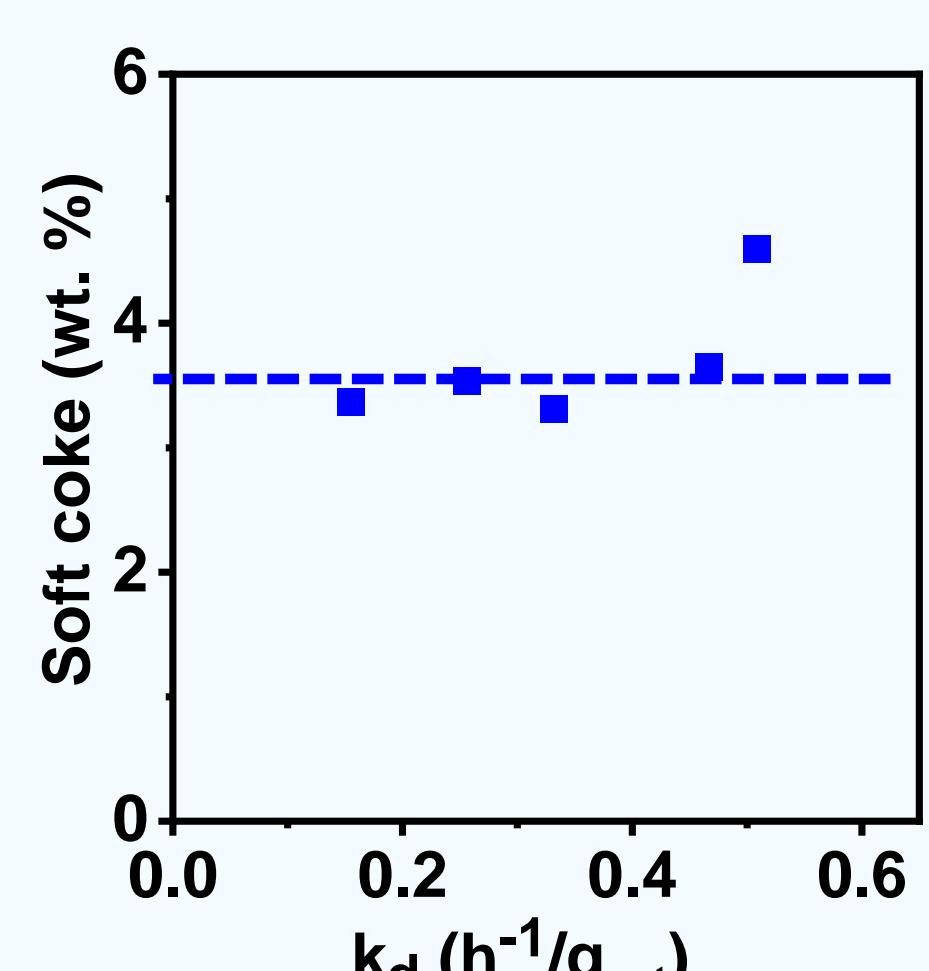


"soft" coke = non toxic coke

- Non toxic type of coke



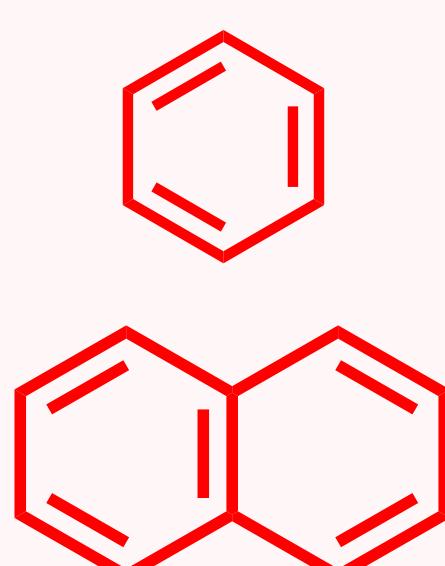
Spectator



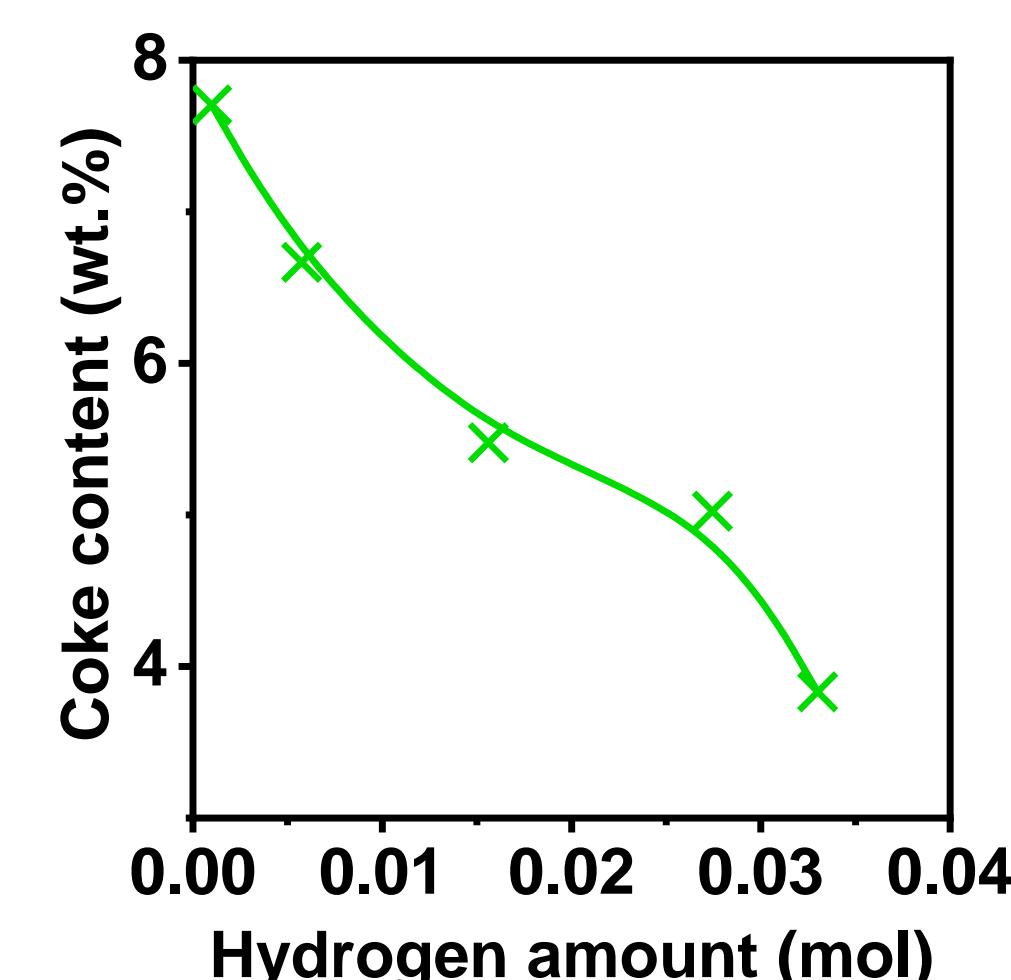
"hard" coke = toxic coke

- Toxic type of coke → inhibit the active species

Actor

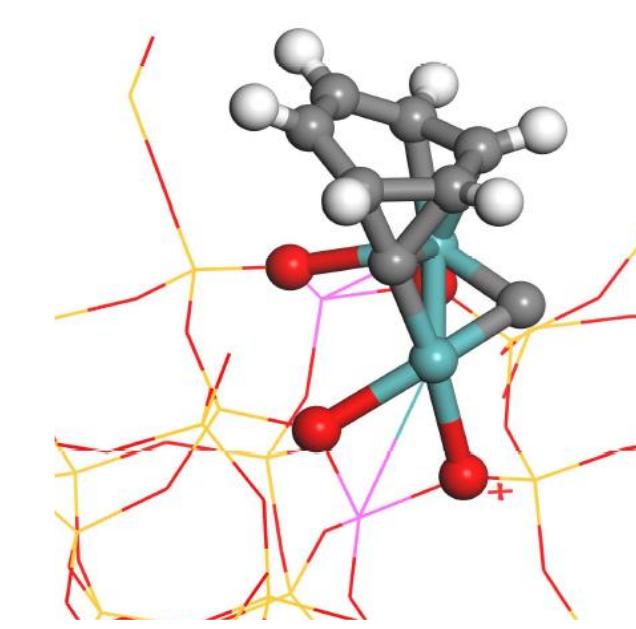


Hydrogen production

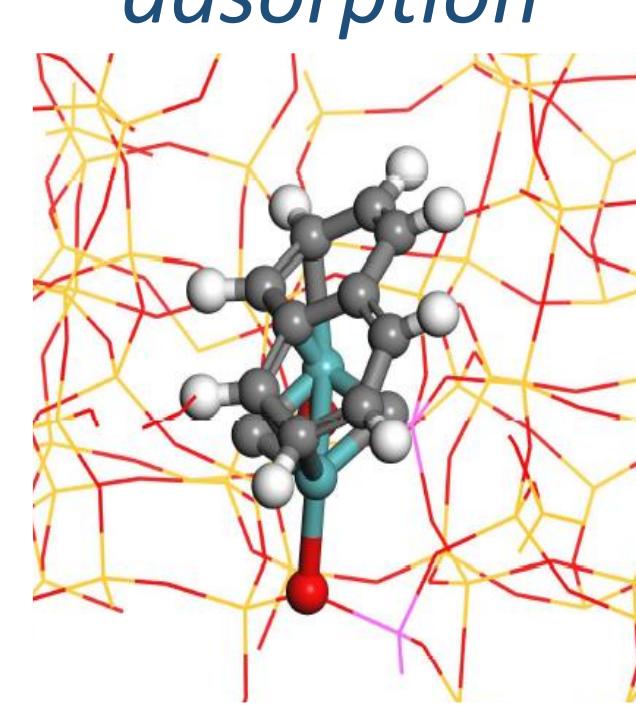


- The higher the H₂ production, the lower the total coke amount

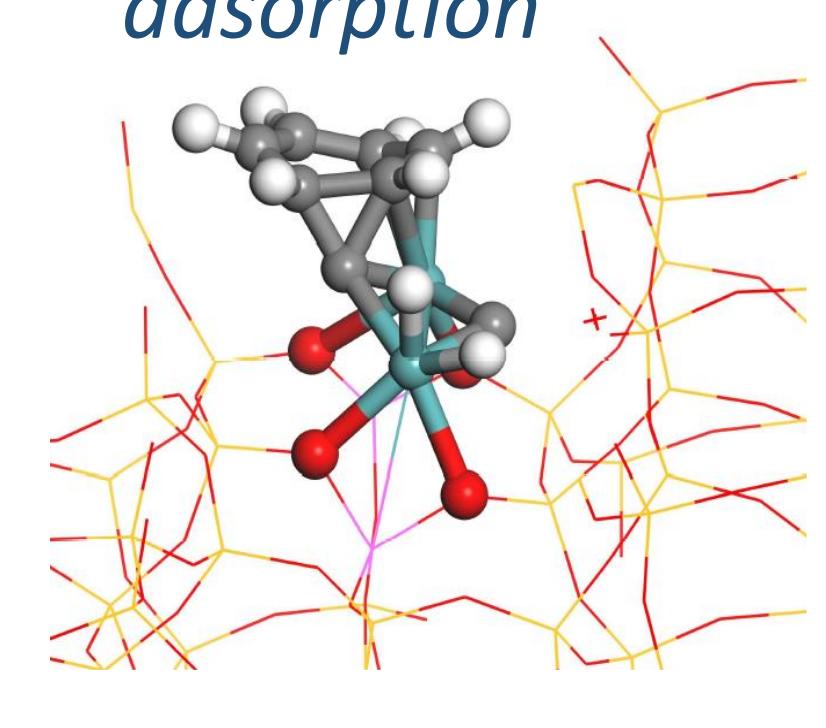
Benzene adsorption



Naphthalene adsorption



CO-hydrogen/benzene adsorption



Presence of hydrogen reduces strength of the benzene adsorption on active sites

Conclusion

- Contact time was subsequently adjusted and methane dehydroaromatization was carried out. The catalyst deactivates in about 10 hours
- Two different type of coke were identified thanks to TGA/TDA analysis: « soft » and « hard » coke
- Soft coke is considered as a non toxic type of coke whereas « hard » coke inhibits the active species
- Autogenous hydrogen prevents the formation of "hard" coke, which is the cause of deactivation

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